Amendments to the Specification

Paragraph beginning on page 15, line 17 and ending on page 16, line 15 is replaced with the following rewritten paragraph.

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In step 1, a catalysis solution is first formed from water, an acid, and an oxidizer. The acid is preferably hydrochloric acid HC1, but other acids may be used, such as sulfuric acid H_2SO_4 , nitric acid HNO03, perchloric acid HClO04, phosphoric acid H3PO4, acetic acid CH₃COOH, formic acid HCOOH, tartaric acid C₄H₆O₆, methanesulfonic acid CH_4SO_3 , ethylsulfonic acid $C_2H_7SO_3$, 4toluenesulfonic acid C7H8SO3, and camphorsulfonic acid (CSA). The oxidizer is preferably ammonium peroxydisulfate (NH₄)₂S₂O₈, but other oxidizers may be used, such as iron chloride FeCl3 and other peroxydisulfate derivates such as $Na_2S_2O_8$ and $K_2S_2O_8$. In step 2, a monomer solution is formed from a solution of a nonconducting monomer and an organic solvent. In the preferred form, the monomer is aniline, but other carbon-based organic monomers can be used, such as pyrrole, thiophene, toluidine, anisidine and other derivatives of aniline such as methylaniline, ethylaniline, 2alkoxyaniline, and 2,5-dialkoxyaniline monomers, for forming polyaniline, polypyrrole, polythiophene, polytoluidine, polyanisidine, polymethylaniline, polyethylaniline, poly(2alkoxyanilines) and poly (2,5-dialkoxyanilines) respectively. The organic solvent is preferably carbon tetrachloride (CCl4), but other organic solvents may be used, such as benzene, toluene, chloroform, methylene chloride, xylene, hexane, diethylether, dichloromethane and carbon disufide. In the preferred form, aniline monomers are dissolved in carbon tetrachloride (CCl₄).